### **CODE- AS 2531**

## B. Pharam. VII SEMESTER EXAMINATION, 2013

### Pharmaceutical Chemistry VII (Medicinal Chem -III)

### **SECTION-A**



vi) Structure of Pantothenic acid.



Q1. b) IUPAC name of indomethacin .

is 1-(p chlorobenzoyl)-2-methyl-6-methoxy-3-indoleacetic acid.

- Q1. c) Two name of MAO inhibitors are Tranylcypromine and phenelzine.
- Q1. d) The name of one diabenzodiazepine derivative drug is Clozapine

Q1. e) Applications of the .

- i) Ketamine. As anesthetics and analgesics.
- ii) Diagonostic agents for determination of blood volume and thyroid function.
- iii) Codeine as expectorant and analgesics.

#### Section -B

### (Descriptive questions. Each question carries 14 marks)

14 X 04 = 56

Q2. Define the term sedatives and hypnotics, Classify them. Write the structure, nomenclature and synthesis of allobarbitone,

and hexobarbitone.

Answer:

sedatives refers to a queting effect accompanied by relaxation and rest but not sleep and hypnotic drugs are used to produce drowsiness and help the onset and maintenance of sleep.

Classes of sedatives and hypnotics are:-

- A) Barbiturates
- amobarbital (Amytal)
- pentobarbital (Nembutal)
- secobarbital (Seconal)
- phenobarbital (Luminal)
- Benzodiazepines (trade names)
  - clonazepam (Klonopin N.America; Rivotril Europe, Asia )
  - diazepam (Valium)
  - estazolam (Prosom)
  - flunitrazepam (Rohypnol)
  - lorazepam (Ativan)
  - midazolam (Versed)
  - nitrazepam (Mogadon)
  - oxazepam (Serax)
  - triazolam (Halcion)
  - temazepam (Restoril, Normison, Planum, Tenox, and Temaze)
  - chlordiazepoxide (Librium)
  - alprazolam (Xanax)
- Nonbenzodiazepine "Z-drugs" sedatives
  - eszopiclone (Lunesta)
  - zaleplon (Sonata)
  - zolpidem (Ambien)
  - zopiclone (Imovane, Zimovane)
- Antihistamines
  - diphenhydramine
  - dimenhydrinate
  - doxylamine
  - mirtazapine
  - promethazine
- Herbal sedatives

- Duboisia hopwoodii
- Chamomile
- Prostanthera striatiflora
- catnip
- kava (Piper methysticum)
- valerian
- cannabis
- passiflora spp.(passiflora incarnata)
- Validol

## Allobarbitone,

Structure :-



IUPAC Name:

5,5-diprop-2-enyl-1,3-diazinane-2,4,6-trione

Synthesis:



## Hexobarbitone.



Synthesis:



Q3. What do you understand by convulsions, classify the anticonvulsant drugs. Write the structure, nomenclature and synthesis

of felbamate, gabapentin and lamotrigine.

Ans: The epilepsies or convulsions are a group of disorders characterized by chronic, recurrent, paroxymal changes in neurologic function caused by abnormalities in electrical activity of the brain.

Classes of the anticonvulsant drugs are

1)Hydantoin derivatives : Phenytoin, Ethantoin.

2)Barbiturate: Phenobarbitone,

3)Oxazolidinedione: Trimethadione, Paramethadione.

4)Succimide derivatives: Phensuccimide, ethosuccimide.

5)Miscellaneous: Gabapentin, felbamate, valproic acid and lamotrigine

Felbamate:-

(3-carbamoyloxy-2-phenylpropyl) carbamate

Gabapentin:

Structure:



Nomenclature:

2-[1-(aminomethyl)cyclohexyl]acetic acid

Synthesis: Cyclohexane 1,1 diacetic acid is esterified with methanol and reacted with ethyl chloroformate to yield 1isocynatomethyl derivative of monoester. This compound is converted to the 1-(aminoethyl) product and the lactam, through the cyclization of the ester and the free amine .The mixture is refluxed with dilute HCl to give Gabapentin.

### Lamotrigine:



Synthesis:-

Reaction of 2,3dichlorobenzoyl chloride with cynide ion form the benzoylcynide derivative that is then treated with aminoguanidine to form Schiff base through loss of water between the carbonyl group and primary amine function of guanidine.Ring closure by addition of the lone free amino group,remaining on the guanidine moiety to the nitrile function is accomplished by base catalysis and yield the product.

Q4. What are Neuroleptics. Enumerate the various phenothiazine derivatives. Write the structure, nomenclature and synthesis of

droperidol and acetophenazine.

Ans: Drugs which use for management and treatment of psychosis and neuroses called neuroleptics.

Phenothiazine derivatives are given in the table

Group	Example	Sedative
<u>Aliphatic</u> compounds	Chlorpromazine (marketed as Thorazine, Chlor-PZ, Klorazine, Promachlor, Promapar, Sonazine, Chlorprom, Chlor- Promanyl, Largactil)	strong
	Promazine (trade name Sparine)	moderate
	<u>Triflupromazine</u> (trade names Clinazine, Novaflurazine, Pentazine, Terfluzine, Triflurin, Vesprin)	strong
	Levomepromazine in Germany and <u>methotrimeprazine</u> in America (trade names Nozinan, Levoprome)	extremely strong
Piperidines	Mesoridazine (trade name Serentil)	strong

	<u>Thioridazine</u> (trade names Mellaril, Novoridazine, Thioril)	strong
	Fluphenazine (trade names Prolixin, Permitil, Modecate, Moditen)	weak/moderate
<u>Piperazines</u>	Perphenazine (sold as Trilafon, Etrafon, Triavil, Phenazine)	weak/moderate
	Prochlorperazine (trade names Compazine, Stemetil)	
	Trifluoperazine (trade name Stelazine)	moderate

### Acetophenazine

бн

1-[10-[3-[4-(2-hydroxyethyl)piperazin-1-yl]propyl]-10H-phenothiazin-2-

yl]ethanone



Droperidol





Q5. Write the classes of antidepressant drugs. Give structure, nomenclature and synthesis of bupropion, despiramine and

doxepin.

Antidepressants are drugs used for the treatment of major depressive disorder and other conditions, including dysthymia, anxiety disorders, obsessive compulsive disorder, eating disorders, chronic pain, neuropathic pain and, in some cases, dysmenorrhoea, snoring, migraines, attention-deficit hyperactivity disorder (ADHD), substance abuse and sleep disorders. They can be used alone or in combination with other medications.

The most important classes of antidepressants are the selective serotonin reuptake inhibitors (SSRIs), serotoninnorepinephrine reuptake inhibitors (SNRIs), tricyclic antidepressants (TCAs) and monoamine oxidase inhibitors (MAOIs). drugs Other used or proposed for the treatment of depression include buprenorphine, tryptophan, low-dose antipsychotics, and St John's wort

**Bupropion**:



±)-2-(tert-Butylamino)-1-(3-chlorophenyl)propan-1-one



Despiramine



3-(10,11-dihydro-5*H*-dibenzo[*b*,*f*]azepin-5-yl)-*N*-methylpropan-1-amine

Synthesis:



Doxepin.



Name: -3-(dibenzo[b,e]oxepin-11(6H)-ylidene)-N,N-dimethylpropan-1-amin

Synthesis:



Q6. Define radiopharmaceuticals. Write Pharmaceutical applications of Au, I and Co. radioisotopes.

Ans: Many of the chemical elements have a number of isotopes. The isotopes of an element have the same number of protons in their atoms (atomic number) but different masses due to different numbers of neutrons. In an atom in the neutral state, the number of external electrons also equals the atomic number. These electrons determine the chemistry of the atom. The atomic mass is the

sum of the protons and neutrons. There are 82 stable elements and about 275 stable isotopes of these elements.

When a combination of neutrons and protons, which does not already exist in nature, is produced artificially, the atom will be unstable and is called a radioactive isotope or radioisotope. There are also a number of unstable natural isotopes arising from the decay of primordial uranium and thorium. Overall there are some 1800 radioisotopes.

Radiopharmaceutical is a preparation intended for in vivo use, that contains radionuclide and use for diagnostic and therapeutic purpose.

At present there are up to 200 radioisotopes used on a regular basis, and most must be produced artificially.

Radioisotopes can be manufactured in several ways. The most common is by neutron activation in a nuclear reactor. This involves the capture of a neutron by the nucleus of an atom resulting in an excess of neutrons (neutron rich).

Some radioisotopes are manufactured in a cyclotron in which protons are introduced to the nucleus resulting in a deficiency of neutrons (proton rich).

The nucleus of a radioisotope usually becomes stable by emitting an alpha and/or beta particle. These particles may be accompanied by the emission of energy in the form of electromagnetic radiation known as gamma rays. This process is known as radioactive decay.

Radioisotopes have very useful properties: radioactive emissions are easily detected and can be tracked until they disappear leaving no trace. Alpha, beta and gamma radiation, like x-rays, can penetrate seemingly solid objects, but are gradually absorbed by them. The extent of penetration depends upon several factors including the energy of the radiation, the mass of the particle and the density of the solid. These properties lead to many applications for radioisotopes in the scientific, medical, forensic and industrial fields

Carbon-14: Used to measure the age of water (up to 50,000 years)

Chlorine-36: Used to measure sources of chloride and the age of water (up to 2 million years) Lead-210: Used to date layers of sand and soil up to 80 years

Tritium (H-3): Used to measure 'young' groundwater (up to 30 years)

# Artificially-produced radioisotopes:

Americium-241:

Used in backscatter gauges, smoke detectors, fill height detectors and in measuring ash content of coal.

Caesium-137:

Used for radiotracer technique for identification of sources of soil erosion and deposition, in density and fill height level switches.

Chromium 57:

Used to label sand to study coastal erosion.

Cobalt-60, Lanthanum-140, Scandium-46, Silver-110m, Gold-198: Used together in blast furnaces to determine resident times and to quantify yields to measure the furnace performance.

# Cobalt-60:

Used for gamma sterilisation, industrial radiography, density and fill height switches.

Gold-198 & Technetium-99m:

Used to study sewage and liquid waste movements, as well as tracing factory waste causing ocean pollution, and to trace sand movement in river beds and ocean floors.

Gold-198: Used to label sand to study coastal erosion.

Hydrogen-3 (Tritiated Water): Used as a tracer to study sewage and liquid wastes

Iridium-192 Used in gamma radiography to locate flaws in metal components.

Krypton-85: Used for industrial gauging.

Manganese-54: Used to predict the behaviour of heavy metal components in effluents from mining waste water.

Nickel-63

Used in light sensors in cameras and plasma display, also electronic discharge prevention and in electron capture detectors for thickness gauges.

Selenium-75: Used in gamma radiography and non-destructive testing.

Strontium-90: Used for industrial gauging.

Thallium-204: Used for industrial gauging.

Ytterbium-169: Used in gamma radiography and non-destructive testing.

# Zinc-65:

Used to predict the behaviour of heavy metal components in effluents from mining waste wate

Q7. Classify vitamins. Write structure , sources and applications of vitamin A and vitamin D.

Each vitamin is typically used in multiple reactions, and, therefore, most have multiple functions.

Classes of vitamins, sources and applications of vitamin A and vitamin D are shown in the table given below:-

Vitamin generic descript or name	<u>Vitamer</u> chemical name(s) (list not complete)	<u>Solubili</u> <u>ty</u>	Recommend ed dietary allowances (male, age 19–70) <sup>[7]</sup>	Deficiency disease	Upper Intake Level (UL/day ) <sup>[기</sup>	Overdose disease	Food sources
<u>Vitamin K</u>	phylloquinone,menaquinones	Fat	120 µg	Bleeding diathesis	N/D	Increases coagulation in patients taking <u>warfarin</u> . <sup>[1]</sup> 9]	Leafy green vegetable s such as spinach, egg yolks, liver
<u>Vitamin E</u>	Tocopherols,tocotrienols	Fat	15.0 mg	Deficiency is very rare; mild <u>hemolytic</u> <u>anemia</u> in newborn infants. <sup>[17]</sup>	1,000 m g	Increased congestive heart failure seen in one large randomized study. <sup>[18]</sup>	Many fruits and vegetable s, nuts and seeds
<u>Vitamin D</u>	Cholecalciferol,Ergocalciferol	Fat	10 µg <sup>(16)</sup>	<u>Rickets</u> and <u>Osteomal</u> acia	50 µg	<u>Hypervitaminosi</u> <u>s D</u>	Fish, eggs, liver, mushroo ms
<u>Vitamin C</u>	Ascorbic acid	Water	90.0 mg	<u>Scurvy</u>	2,000 m g	<u>Vitamin C</u> megadosage	Many fruits and vegetable s, liver

Vitamin generic descript or name	<u>Vitamer</u> chemical name(s) (list not complete)	<u>Solubili</u> <u>ty</u>	Recommend ed dietary allowances (male, age 19–70) <sup>[7]</sup>	Deficiency disease	Upper Intake Level (UL/day ) <sup>[기</sup>	Overdose disease	Food sources
<u>Vitamin B<sub>9</sub></u>	<u>Folic acid, folinic acid</u>	Water	400 µg	Megaloblastic anemiaand Deficiency during pregnancy is associated with <u>birth</u> <u>defects</u> , such as <u>neural</u> <u>tube</u> defects	1,000 μ g	May mask symptoms of vitamin B <sub>12</sub> deficiency; <u>o</u> <u>ther effects</u> .	Leafy vegetable s, pasta, bread, cereal, liver
<u>Vitamin B<sub>7</sub></u>	Biotin	Water	30.0 µg	<u>Dermatitis, enteritis</u>	N/D		Raw egg yolk, liver, peanuts, certain vegetable
<u>Vitamin B<sub>6</sub></u>	Pyridoxine.pyridoxamine.pyridoxal	Water	1.3–1.7 mg	<u>Anemia<sup>[14]</sup> peripheral</u> neuropathy.	100 mg	Impairment of <u>proprioception</u> , nerve damage (doses > 100 mg/day)	Meat, vegetable s, tree nuts, bananas
<u>Vitamin B₅</u>	Pantothenic acid	Water	5.0 mg <sup>[12]</sup>	Paresthesia	N/D	Diarrhea; possibly nausea and heartburn. <sup>[13]</sup>	Meat, broccoli, avocados
<u>Vitamin B<sub>3</sub></u>	<u>Niacin, niacinamide</u>	Water	16.0 mg	<u>Pellagra</u>	35.0 mg	Liver damage (doses > 2g/day) <sup>[11]</sup> and <u>ot</u> <u>her problems</u>	Meat, fish, eggs, many vegetable s, mushroo ms, tree nuts
<u>Vitamin B<sub>2</sub></u>	Riboflavin	Water	1.3 mg	Ariboflavinosis	N/D		Dairy products, bananas, popcorn, green beans, asparagu s

Vitamin generic descript or name	<u>Vitamer</u> chemical name(s) (list not complete)	<u>Solubili</u> <u>ty</u>	Recommend ed dietary allowances (male, age 19–70) <sup>[2]</sup>	Deficiency disease	Upper Intake Level (UL/day ) <sup>[2]</sup>	Overdose disease	Food sources
<u>Vitamin B<sub>12</sub></u>	<u>Cyanocobalamin,hydroxycobalamin,methylc</u> obalamin	Water	2.4 µg	Megaloblastic anemia <sup>[15]</sup>	N/D	Acne-like rash [causality is not conclusively established].	Meat and other animal products
<u>Vitamin B<sub>1</sub></u>	Thiamine	Water	1.2 mg	<u>Beriberi, Wernicke-</u> Korsakoff syndrome	N/D <sup>191</sup>	Drowsiness or muscle relaxation with large doses. <sup>[10]</sup>	Pork, oatmeal, brown rice, vegetable s, potatoes, liver, eggs
<u>Vitamin A</u>	<u>Retinol, retinal,</u> and four <u>carotenoids</u> including <u>beta carotene</u>	Fat	900 hđ	<u>Night-</u> <u>blindness,Hyperkerat</u> osis, and <u>Keratomalacia<sup>(8)</sup></u>	3,000 µ g	<u>Hypervitaminosi</u> <u>s A</u>	Orange, ripe yellow fruits, leafy vegetable s, carrots, pumpkin, squash, spinach, liver, soy milk, milk

Vitamin	Food source
Vitamin B₁ ( <u>Thiamine</u> )	Rice bran
Vitamin A ( <u>Retinol</u> )	Cod liver oil
Vitamin C ( <u>Ascorbic acid</u> )	<u>Citrus</u> , most fresh foods
<u>Vitamin D</u> (Calciferol)	Cod liver oil

Vitamin	Food source
Vitamin B <sub>2</sub> ( <u>Riboflavin</u> )	<u>Meat, eggs</u>
( <u>Vitamin E</u> ) ( <u>Tocopherol</u> )	Wheat germ oil, unrefined vegetable oils
<u>Vitamin B<sub>12</sub> (Cobalamins)</u>	liver,eggs, animal products
Vitamin K <sub>1</sub> ( <u>Phylloquinone</u> )	Leafy green vegetables
Vitamin B₅ ( <u>Pantothenic acid</u> )	<u>Meat, whole grains,</u> in many foods
Vitamin B <sub>7</sub> ( <u>Biotin</u> )	Meat, dairy products, eggs
Vitamin B <sub>6</sub> ( <u>Pyridoxine</u> )	Meat, dairy products
Vitamin B₃ ( <u>Niacin</u> )	<u>Meat, eggs, grains</u>
Vitamin B <sub>9</sub> (Folic acid)	Leafy green vegetables

Structure of Vitamin – A is



Structure of vitamin D is

